SUDDEN SOLVENT EVAPORATION IN BUBBLE ELECTROSPINNING FOR FABRICATION OF UNSMOOTH NANOFIBERS

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Solvent fraction in a multiple solvent system plays a great important for controlling fiber morphology in the bubble electrospinning. The sudden release of air pressure when a polymer bubble is broken and an extremely high velocity of jets are two main factors for solvent evaporation, which can be used for fabrication of unsmooth fibers.

Key words: composite, nanofiber, bubble electrospinning, unsmooth structure, diameter, water contact angle

Introduction

Nanoscale porous fibers have much more applications than smooth nanofibers. In this paper we will study a multiple solvent system in the bubble electrospinning for fabrication of PS/PVP composite nanofibers, PS (polystyrene) is an amorphous polymer and has good stability [1], and it can be easily dissolved in dimethyl formamide (DMF) [2], and PVP (polyvinylpyrrolidone) is a newly developed material which is a compound of PVC (polyvinylchloride) and PP (polypropylene) [3, 4], the former has excellent hydrophilic while the latter has good flexibility [5].

Experimental

We prepared six samples for spun solution. The samples A and B were, respectively, 20 wt.% and 17.9 wt.% PS solutions with DMF solvent. The sample C was 27.27 wt.% polyvinilpyrrolidone (PVP) solution with C₂H₅OH as solvent; The sample D was prepared by dropping sample C into sample B gradually, till the resultant solution became agglomerated, see fig. 1(b). The sample E was 20 wt.% PS/PVP (mass mixing ratio of 7:3) with DMF as solvent. The sample F was prepared by dropping C₂H₅OH into sample E till the solution became agglomerated, fig. 1(c). The spinning process by the bubble electrospinning was illustrated in fig. 2, where the voltage was kept as 20 kV and the receipt distance was 13 cm.

Solvent evaporation

Sudden solvent evaporation occurs when air pressure decreases greatly and suddenly. During the bubble electrospinning or bubbl spinning [3-8], an air-flow is input into the spun solution with a high pressure, and a polymer bubble is formed on the surface. When the bubble is broken, a sudden pressure drop is happened, and solvent evaporation occurs which

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makes each debris lose solvents for a fast solidification. This evaporation process is similar to that in flash evaporation [9], and sometimes the solvents can be boiled due to the sudden pressure release, making a fast solvent evaporation. Additionally, due to high ejecting velocity of each debris, which is as high as 300 m/s, the air drag pushes the debris to form a fiber, and the surface tension makes the fiber cylindrical. According to Bernoulli equation:

\[ \frac{1}{2} u^2 + \frac{P}{\rho} = B \]  

where \( u \) is the velocity of the moving jet, \( P \) – the fluid pressure, \( \rho \) – the density, \( B \) – the Bernoulli constant. A debris from the broken bubble is accelerated from almost zero velocity to as high as 300 m/s, according to eq. (1), a great pressure drop occurs when the debris is ejected, this results in a second solvent evaporation in the spinning process. A spun solution with multiple fractions including solvents always leads to an extremely fast solvent evaporation, resulting in unsmooth surface of the obtained fibers.

**Discussion and conclusion**

Figure 3 is scanning electron microscope (SEM) of nanofibers from the sample A, and nanoscale pores are observed on fibers surface. Figure 4 is SEM of nanofibers from the sample D (PVP + C₂H₅OH). Figure 5 is SEM of nanofiber about liquid F.

Figure 1. Spun solutions with different fractions of components and solvents

Figure 2. The bubble electrospinning
Figure 3(b) showed that the surface of PS nanofiber was porous nanostructure and rough, but fig. 4(b) displayed that the holes on the surface of PS/PVP composite nanofiber reduced obviously, and the rough degree of the surface was lowered clearly, while the smooth performance was improved. Although the components of nanofiber about fig. 5(b) were the same to that about fig. 4(b), there is a clear hierarchy structure on the surface of PS/PVP composite nanofiber. From fig. 5(b) we can see that the roughness of nanofiber surface is higher, there are more holes on the nanofiber surface, and even there are deep troughs on the nanofiber surface.

The components of nanofibers on figs. 4 and 5 are the same, but there are big differences about the surface morphology, this showed that the order ratio of solute and solvent in the solution have an effect on the hierarchy structure about the nanofiber surface [9, 10]. In fig. 4 we prepare the spinning solution on PS and PVP respectively, so even if when they are mixed their own dissolution was not destroyed, also good for bubble electrospinning, PVP can fill up the holes on the surface of PS nanofibers, so the holes on the surface of PVP nanofibers were reduced and it improve the smoothing effect of the surface. In fig. 5 we first let PVP and PS dissolve in DMF together. Because the ethanol and DMF are immiscible, so when the ethanol dropped into the aforementioned solution it had a greater influence on composite solution, which makes the surface of nanofiber appear unsmooth structure during the spinning process.

Figure 4. Composite nanofiber of PS/PVP(PS/DMF+PVP/C₂H₅OH)

Figure 5. Composite nanofiber of PS/PVP(PS/PVP/DMF+C₂H₅OH)
The diameter distribution of nanofibers in figs. 3-5, were shown in figs. 6(a)-(c), respectively. From fig. 6(a) we can find that the diameter of PS nanofibers range widely from 260 nm to 460 nm and its diameter concentrate between 310 nm and 360 nm. It can be seen from fig. 6(b) that during the same spinning technique, the diameter of PS/PVP composite nanofibers range narrowly from 40 nm to 350 nm, the diameter of most nanofibers concentrate between 50 nm to 130 nm. Although the fiber content of fig. 6(c) is same with that of fig. 6(b), due to the different preparation of spinning solution, we can find that the diameter of nanofibers about fig. 6(c) has changed from 40 nm to 540 nm, the diameter of nanofibers concentrate between 90 nm and 230 nm.

To discuss the effect of nanofiber surface unsmooth structure on hydrophilic performance, we test three water contact angles of the surface about those three kinds of nanofiber membranes. Water contact angle of nanofiber shown in figs. 3-5 was shown in fig. 7. From fig. 7 we can see that in fig. 3 the PS nanofiber membrane surface water contact angle is 98°, in fig. 4 the PS/PVP composite nanofiber membrane surface contact angle is 108°, and in fig. 5 in the the PS/PVP composite nanofiber membrane surface contact angle is 83°, this hydrophilic differences is because of the nanofiber surface structure, the nanofiber surface in fig. 5 there has been a lot of grooves which make the fiber surface occurs capillary effect and improve the hydrophilicity of the nanofiber membranes.

In order to discuss the effect of fiber surface structure which was prepared by the new bubble electrospinning technology and spinning process on the nanomembrane tensile property, we respectively tested the tensile strength and tensile elongation rate of the aforementioned three nanomembranes, as shown in fig. 8. From fig. 8 we can see that alt-
though the fiber components of the nanomembranes b and c were the same, nanomembrane c has the higher tensile strength and the smaller tensile elongation rate. This is because the high rough degree on the nanomembrane c surface increases fiber slip resistance [11].

![Figure 7. Water contact angle of nanomembrane in figs. 3-5](image1)

![Figure 8. Tensile property of nanomembrane in figs. 3-5](image2)

The surface average pore size and air velocity of nanomembrane a, b, and c are shown in fig. 9. It can be seen in fig. 9 that the variation trend of the surface mean pore size keep consistent with that of the air velocity, and the nanomembrane c has the maximum surface average pore size and air velocity. Its main reasons were the un-smooth property on the surface and the more rough fiber diameter greatly increased opportunities of producing gap on the nanomembrane surface [12].

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